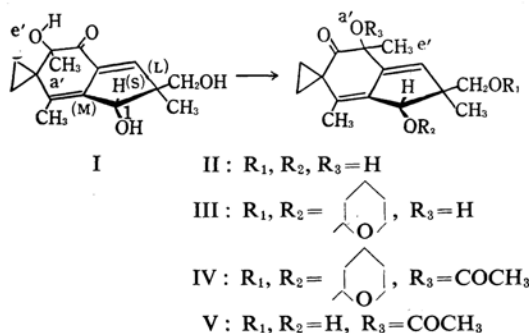


## The Stereochemistry of Illudin S (*Lampterol*)<sup>1, 2a)</sup>

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Recently the structure of illudin S,<sup>3)</sup> an antitumor substance obtained from a culture liquid of *Clitocybe illudens*, has been proposed. We now wish to propose stereochemical formula I for illudin S on the basis of the following evidence.



The infrared spectrum<sup>4)</sup> of illudin S in carbon tetrachloride ( $7.84 \times 10^{-4}$  mol.) showed absorption bands at  $3631 \text{ cm}^{-1}$  ( $\epsilon$ , 58.4, free primary OH),  $3601 \text{ cm}^{-1}$  ( $\epsilon$ , 68.4 free secondary allylic OH) and  $3485 \text{ cm}^{-1}$  ( $\epsilon$ , 67.1 bonded tertiary OH), while illudin M,<sup>3)</sup> in which the hydroxymethyl group is replaced by a methyl group, showed bands with almost identical intensities at  $3601 \text{ cm}^{-1}$  ( $\epsilon$ , 61.6, free secondary

OH) and  $3485 \text{ cm}^{-1}$  ( $\epsilon$ , 71.7, bonded tertiary OH) in carbon tetrachloride ( $7.72 \times 10^{-4}$  mol.).

These observations clearly indicate the trans orientation of the hydroxymethyl group and the secondary hydroxyl group.

Isoilludin S<sup>2a, b)</sup> (II), formed by the acyloin rearrangement of illudin S with alumina, has infrared absorption bands at  $3632 \text{ cm}^{-1}$  ( $\epsilon$ , 41.4, free primary OH),  $3601 \text{ cm}^{-1}$  ( $\epsilon$ , 56.5, free secondary OH) and  $3503 \text{ cm}^{-1}$  ( $\epsilon$ , 65.2, bonded tertiary OH) in carbon tetrachloride ( $1.03 \times 10^{-3}$  mol.). These data show that the tertiary hydroxyl group has a pseudoequatorial conformation in illudin S (the stronger hydrogen bond) and a pseudoaxial conformation in isoilludin S.

The treatment of isoilludin S (II) with dihydropyran and the subsequent acetylation of the dipyranyl compound III, gave IV, which was then hydrolyzed with methanolic hydrogen chloride to monoacetylisilludin S (V):  $\tau$ , 8.90, 8.42, 8.31 (methyl groups), 7.89 ( $2 \times \text{CH}_3\text{CO}_2-$ ), 6.58 ( $-\text{CH}_2\text{OH}$ ), 5.31 (proton  $\alpha$  to hydroxyl group), 4.62 (olefinic proton), and 7.95~9.15 (two methylene groups of cyclopropane), whose infrared spectrum in carbon tetrachloride ( $1.00 \times 10^{-3}$  mol.) showed, in addition to an absorption band at  $3603 \text{ cm}^{-1}$  resulting from a secondary hydroxyl group, an absorption band at  $2632 \text{ cm}^{-1}$  (free primary OH) with a rather decreased intensity and a newly-appeared band at  $3532 \text{ cm}^{-1}$  (bonded primary OH).

These results provide evidence of the cis orientation of the hydroxymethyl group and of the tertiary hydroxyl groups in isoilludin S (II).

The stereochemistry of illudin S should, therefore, be represented by formula I in view of the known steric course of acyloin migration.<sup>5)</sup>

The absolute configuration of illudin S was determined by applying Horeau's procedure.<sup>7)</sup>

When I ( $0.2 \times 10^{-3}$  mol.) was treated with ( $\pm$ )  $\alpha$ -phenylbutyric anhydride ( $0.1 \times 10^{-2}$  mol.),

1) Presented at the IUPAC Symposium on the Chemistry of Natural products, Kyoto, Japan, April, 1964.

2) a) *Lampterol*, which was isolated from *Lampteromyces Japonicus*, has been proven identical with illudin S on the basis of direct comparison of their nuclear magnetic resonance and infrared spectra as well as mixing melting point.

Cf., H. Shirahama, Y. Fukuoka and T. Matsumoto, *J. Chem. Soc. Japan, Pure Chem. Sec. (Nippon Kagaku Zasshi)*, 83, 1289 (1962); H. Shirahama, Y. Fukuoka and T. Matsumoto, *This Bulletin*, 35, 1047 (1962); K. Nakanishi, M. Ohashi, N. Suzuki, M. Tada, Y. Yamada and S. Inagaki, *Yakugaku Zasshi*, 83, 377 (1963).

b)  $\text{C}_{15}\text{H}_{24}\text{O}_4$ , m. p.  $179\sim 180^\circ\text{C}$ ;  $\lambda_{\text{EtoH}}^{\text{max}}$  252 m $\mu$  ( $\log \epsilon$ , 4.3)

$\nu_{\text{KBr}}^{\text{max}}$  3420, 1660, 1645, 1365, 1018  $\text{cm}^{-1}$ ;  $\tau$ , 8.84, 8.49, 8.32 (three methyl groups), 7.7~8.9 (two methylene groups of cyclopropane), 7.55 (OH), 6.61 ( $\text{CH}_2\text{OH}$ ), 5.40 (proton  $\alpha$  to hydroxyl group), 4.30 (olefinic proton). Details about the structure of isoilludin S will be reported in the near future.

3) T. C. McMorris and M. Anchel, *J. Am. Chem. Soc.*, 85, 831 (1963).

4) The infrared spectra were measured on a Nippon Bunko grating infrared spectrometer, Model DS-401G, using 50 mm. KRS cell under high resolution condition.

5) Nuclear magnetic resonance spectra were measured on JNM-3H-60 high resolution NMR spectrometer.

6) W. A. Jacobs and W. S. Pelletier, *J. Org. Chem.*, 22, 1428 (1957).

7) A. Horeau, *Tetrahedron Letters*, 506 (1961); 965 (1962).

the resultant  $\alpha$ -phenylbutyric acid showed a rotation of  $\alpha_D^{20} + 8.75^\circ$  (optical yield, 9.01%). Since the steric requirement of substituents around  $C_1$  can be regarded to be as indicated in formula I, it may be concluded that illudin S should have the absolute configuration of I.

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